					<i>-</i>	
California State University San Ma	arcos	• NEW COL	URSE •	FORM	CHEM 5	<b>D</b> '
ORIGINATOR'S SECTION:	<b>D</b> 1 1 m	177 07		E II 2000)		_
1. College:	Desired Teri	m and Year of In	nplementation (e.g	,, Fall 2008):		
☐ CHABSS ☐ CoBA ☐ CoEHHS ☒ CSM	Fall 2017					
2.Course is to be considered for G.E.? (If yes, also fill out appropriate GE form*)   Yes   No						
3. Course will be a variable-topics (generic) course?						
4. Course abbreviation and Number:* CHEM 504						
5. Title: (Titles using jargon, slang, copyrighted names, trade names, or any non-essential punctuation may not be used.) Advanced Inorganic Chemistry						
6. Abbreviated Title for PeopleSoft: (no more than 25 characters, including spaces) Adv. Inorganic Chem.						
7. Number of Units: 3						
8. Catalog Description: (Not to exceed 80 words; language should conform to catalog copy. Please consult the catalog for models of style and format; include all necessary information regarding consent for enrollment, pre- and/or corequisites, repeated enrollment, crosslisting, as detailed below. Such information does not count toward the 80-word limit.)						
Surveys of elements and compounds of both the main group and transition series, with an emphasis on rationalizing patterns of structure, stability and reactivity across the periodic table. Applications to catalysis, geochemistry and biochemistry will be introduced. <i>Prerequisite: A minimum grade of C (2.0) in CHEM 404 or classified graduate standing.</i>						
9. Why is this course being propose	ed?					
This course is being proposed as part of the new Masters in Chemistry program. CHEM 504 will serve as a required core course in the option in chemistry.						
10. Mode of Instruction*			-			П
For definitions of the Course Classi http://www.csusm.edu/academic_pi ling/catalogcurricula/DOCUMENT Instructional%20Mode%20Conven	rograms/curri TS/Curricular	culumschedu	Type of Instruction	Number of Credit Units	Instructional Mode (Course Classification Number)	
	The state of the s		Lecture	3	C-02	
			Activity			
			Lab			
11. Grading Method:*  Normal (N) (Allows Letter Grade +/-, and Credit/No Credit)  Normal Plus Report-in-Progress (NP) (Allows Letter Grade +/-, Credit/No Credit, and Report-in-Progress)  Credit/No Credit Only (C)  Credit/No Credit or Report-in-Progress Only (CP)						
12. If the (NP) or (CP) grading system was selected, please explain the need for this grade option.						
13. Course Requires Consent for Enrollment?   Yes   No						

Credit/No Credit or Report-in-Progress Only (CP)	
12. If the (NP) or (CP) grading system was selected, please explain the need for this grade of	otion.
13. Course Requires Consent for Enrollment? ☐ Yes ☒ No	
Faculty Credential Analyst Dean Program/Department - Director/Chair  14. Course Can be Taken for Credit More than Once? Yes No If yes, how many times? (including first offering)	DECEIVE MAR 1.7 2017
15. Is Course Crosslisted:  Yes No	1 1 2 2 1 2 1 1
If yes, indicate which course and check "yes" in item #22 below.	BY:
<b>16. Prerequisite(s):</b> ✓ Yes ☐ No CHEM 404 or classified graduate standing.	
17. Corequisite(s): Yes No	

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18. Documentation attached:	Syllabus 🛭	Detailed Course Outlin	ie		
19. If this course has been offer		se enter topic abbreviati	on, number, and suffix	*	
20. How often will this course b	oe offered once est	ablished?* once every ye	ear or every 3 <sup>rd</sup> semester		
PROGRAM DIRECTOR/CHA	IR - COLLEGE (	CURRICULUM COMM	ITTEE SECTION:		
(Mandatory information – all ite					
<ol><li>Does this course fulfill a rec for a major, majors in other de</li></ol>					
If yes, please specify:  Core course in the Chemistry	option of the Maste	ers of Science in Chemistry	, and an elective in the l	piochemistry option	
22. Does this course impact oth check "yes" and obtain signatur	er discipline(s)? (e.)	(If there is any uncertaint No	y as to whether a partici	ılar discipline is a <u>f</u>	fected,
If yes, obtain signature(s). Any o	bjections should be	e stated in writing and atta	ched to this form.		
Discipline				Support	Oppose
	Signature		Date		
Discipline _				Support	Oppose
	Signature		Date		
GNATURES : (COLLEGE 1	LEVEL):		(UNIVERS	SITY LEVEL)	
Schmidt Originator (please print or type name)	8/4 Date	/2016	5. UCC Committee Chair		Date
Program Director/Char	8/9// Date		6. Vice President for Academic	Affairs (or Designee)	Date
College Curriculum Committee	12/14/1	6	7. Decided (C. Decide)		<u> </u>
Muand Date	12/14/16		7. President (or Designee)		Date
College Dean (or Designee)	Date				
	0-				
Office of Academic Programs	PS Banner:	Catalog	ji	Revised 3/28/2007	

<sup>\*</sup> If Originator is uncertain of this entry, please consult with Program/Department Director/Chair.



## Chemistry 504-Advanced Inorganic Chemistry PROSPECTIVE COURSE OUTLINE

**Dr. Michael H. Schmidt** schmidt@csusm.edu 321 Science Hall 2 760-750-4138

**Course Description:** A survey of elements and compounds of both the main group and transition series, with an emphasis on rationalizing patterns of structure, stability and reactivity across the periodic table. Applications to catalysis, geochemistry and biochemistry will be introduced.

## **Student Learning Outcomes:**

- Students will be able to use patterns of structure and bonding to identify inorganic compounds, complexes or solids that are likely to be stable.
- Students will be able to use patterns of reactivity to predict the outcomes of chemical reactions for elements throughout the periodic table.
- Students will be able to explain, in clear written English, with diagrams or mathematics where appropriate, why certain compounds, complexes or solids are stable, and why there are predictable patterns of reactivity for different elements.
- Students will be able to use diagrams and calculations to predict properties of inorganic compounds, complexes and solids.

**Textbooks:** F. Albert Cotton, Geoffrey Wilkinson, Carlos A. Murillo, Manfred Bochmann, *Advanced Inorganic Chemistry, Sixth Edition.* New York: John Wiley & Sons, Inc., 1999. (CWMB)

CWMB is a daunting book of over 1300 pages. You won't be reading the whole thing, as it's too much information to digest in one semester, even for grad students. Selected sections of the chapters will be assigned which focus on chemistry that is unique or typical for the elements covered in a given week. For example, in Week 14, we may focus on heteropolymolybdates and heteropolytungstates, high oxidation state compounds used in homogeneous catalysis, and multiple metal-metal bonds, as these are some of the most interesting things that elements 40-43 and 72-75 do.

Other Inorganic Textbooks will be on reserve in the library. Students have almost always found that looking at another textbook's treatment of the same subject is very helpful.

**Course Activities:** Students will take bi-weekly quizzes on the lecture and reading material, and a final exam encompassing the entire semester.

Occasional homework will be given that links the material in each chapter of the book to recently published literature. This homework will model, for the student, the sorts of connections that can be made between the material of the course and the literature. This should help them with the term paper described below.

An end-of-semester term paper of roughly 10 pages will involve students applying concepts of the course to published literature in their own field of interest. This assignment fulfills the All-University Writing Requirement.

## **Grading Scheme:**

	# of Items	Pts. Per Item	<b>Total Points</b>
Quizzes	7	20	140
Homework	5	20	100
Term Paper	1	50	50
Final Exam	1	100	100
			390

Anticipated schedule: (subject to change)

Lectures	Topics	Readings in CWMB
Week 1	Review of Bonding and Coordination Chemistry	Chapter 1
Week 2	Hydrogen	Chapter 2
Week 3	Groups 1 & 2	Chapters 3 & 4
Week 4	Boron, Group 13	Chapters 5 & 6
Week 5	Carbon, Group 14	Chapters 7 & 8
Week 6	Nitrogen, Group 15	Chapters 9 & 10
Week 7	Oxygen, Group 16	Chapters 11 & 12
Week 8	Halogens, Noble Gases	Chapters 13 & 14
Week 9	Group 12 (Zn, Cd, Hg)	Chapter 15

Week 10	Transition Metal Chemistry	Chapter 16, 16-1 through 16-7
Week 11	Organometallic Chemistry	Chapter 16, 16-8; Parts of Chap. 21
Week 12	First Transition Series, Ti through Mn	Chapter 17, A-D
Week 13	First Transition Series, Fe through Cu	Chapter 17, E-H
Week 14	Lower Transition Series, Zr through Re	Chapter 18, A-D
Week 15	Lower Transition Series, Ru through Au	Chapter 18, E-I